

ISOLATION AND CHARACTERIZATION OF STEROLS AND AN ALIPHATIC ALCOHOL FROM *TSUGA DUMOSA* D. DON OF NEPAL

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Abstract: In the course of our continuing research on Himalayan conifers, the phytochemical screening and isolation of chemical constituents of *Tsuga dumosa* D. Don of Nepali origin was carried out. Three sterols, namely $\hat{\alpha}$ -sitosterol (1), stigmasterol (2), $\hat{\alpha}$ -sitosterol glucoside (3) and an alcohol, nonacosan-10-ol (4) were isolated. Their structures were elucidated on the basis of spectroscopic evidences.

Key Words: Phytochemical screening; *Tsuga dumosa*; Sterols; Nonacosan-10-ol.

INTRODUCTION

Tsuga dumosa D. Don is an economically as well as medicinally important conifer belonging to family Pinaceae.^{1,2} It is commonly known as “Hemlock Spruce” and locally called as “Dhupi” or “Thingre Salla”. In Nepal, it is found in the northern belt mainly in the damp and adverse areas of Sagarmatha National Park at an altitude of 2100–3600 m in association with other conifers like *Taxus*, *Abies* and *Picea*.³ The plant has been extensively used for timbering and lumber products because of its resistance to decay. The bark of this plant is a rich source of tannin, hence can be used for dyeing.⁴ A solid self bio-adhesive herbal compound isolated from *T. dumosa* was used for the tropical treatment of the oral mucosal disorder and for patients exhibiting herpetic stomatitis lesions, aphthous ulcers, mucosal inflammation, toothache and lesions on the lips and gingival.⁵ Among the isolated compounds from *T. dumosa* include lignans, sesquiliglan, dumosaol, along with some other lignans.^{6,7} In the course of our continuing research on Himalayan conifers,⁸ the present study has resulted on isolation of $\hat{\alpha}$ -sitosterol, stigmasterol, $\hat{\alpha}$ -sitosterol glucoside and nonacosan-10-ol from *Tsuga dumosa* D. Don (**Fig. 1**). The structures of the isolated compounds were assigned mainly on the basis of their spectral data.

MATERIALS AND METHODS

General Experimental Procedure: Melting point was measured on electronic melting point apparatus (Ogawa Seiki Co. LTD, Tokyo, Japan) and was uncorrected. IR spectra were obtained on neat using Perkin Emler 1310 Infrared Spectrophotometer. Mass spectra were carried out on JEOL MSRoute. ¹H-NMR and ¹³C-NMR spectra were obtained on

Bruker WM400 NMR spectrometer. Thin layer chromatography (TLC) was carried out on silica gel G pre-coated plates. Separation and purification were performed by column chromatography on silica gel (60-120 mesh)

Plant Material: The fresh twigs of *T. dumosa* were collected from the Nammo-Bauddha area of Kavrepalanchok district during August, 2004 and were identified by Prof. Dr. R. P. Chaudhary, Professor of Central Department of Botany, Tribhuvan University, Kirtipur, Kathmandu, Nepal. Voucher specimen was deposited at National Herbarium and Plant Laboratories, Godawari, Lalitpur, Nepal.

Extraction and Isolation: The air dried twigs of *T. dumosa* (1.0 Kg) were extracted with 5 L ethyl alcohol (EtOH) (90% × 3) at around 60°C for 6 hrs. The crude EtOH extract was evaporated in vacuum. The residue was subjected to fractionation by using hexane, chloroform, n-butyl alcohol to get respective fractions. Hexane and chloroform fractions were subjected to column chromatography on silica gel, eluted with different polarity. From hexane fraction compound **1** (7.0 mg), **2** (12.0 mg), and **4** (38.0 mg) were isolated and compound **3** (16.0 mg) was isolated from chloroform fraction (**Fig. 1**).

RESULTS AND DISCUSSION

Phytochemical screening of *T. dumosa* showed the presence of volatile oil, fatty acids, triterpenes, sterols, carotenoids, polyphenols, reducing compounds, anthracenocoids, coumarins, flavone aglycones, tannins and saponins. In the light of the phytochemical screening, we decided to examine hexane and chloroform fractions for finding bioactive compounds and therefore these fractions were subjected to column chromatography on silica gel, eluted with solvent of

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increasing polarity. From hexane fraction compound **1** (7.0 mg), **2** (12.0 mg), and **4** (38.0 mg) were isolated and compound **3** (16.0 mg) was isolated from chloroform fraction (Fig. 1).

Compound **1** was a crystalline, melted at 135°C. It gave positive Libermann–Burchard test with greenish red color indicating the compound to be sterol. The mass spectrum showed M^+ at m/z at 414 corresponding to the molecular formula $C_{29}H_{50}O$. The IR spectrum showed presence of –OH group at 3400 cm^{-1} and a peak at 1063 cm^{-1} corresponding to –C–O group. Two singlets at δ 1.02 and at 0.66 are methyl group of C-19 and C-18 respectively. The doublets at δ 0.93 (d, $J = 6.1\text{ Hz}$); 0.81 (d, $J = 6.9\text{ Hz}$); 0.82 (d, $J = 6.9\text{ Hz}$) and 0.84 (t, $J = 7.3\text{ Hz}$) were accounted for methyl group at C-21, C-25, C-27 and C-29. A signal at δ 5.33 in $^1\text{H-NMR}$ can be accounted for an olefinic proton H-6 and a multiplet at δ 3.50 equivalent to a proton was assigned for the proton H-3. The low field signal of H-3 due to the attachment of $\hat{\alpha}$ -OH group at C-3 carbon. Thus, the assignment of hydroxyl group at C-3 and the double bond at C-5 were assigned accordingly. Three multiplets equivalent to two protons each appeared at δ 1.82, 1.97 and 2.26 were assigned for three CH_2 groups. The remaining protons have appeared as multiplets at δ 1.05–1.54. From IR spectroscopy, mass and NMR spectroscopic data and comparison with reported spectral data and comparison of TLC with authentic sample, the compound was assigned as $\hat{\alpha}$ -sitosterol.

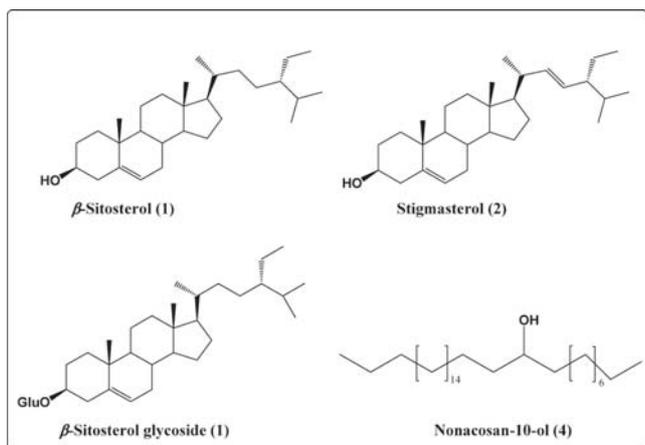


Fig.1: Structure of isolated compound

The compound **2** was white crystalline having melting point 165°C. It gave positive Libermann–Burchard test with greenish red color indicating the compound to be sterol. The mass spectrum showed M^+ at m/z at 412 corresponding to the molecular formula $C_{29}H_{48}O$. The IR spectrum showed the presence of –OH group (3350 cm^{-1}) and a peak at 1063 cm^{-1} corresponding to –C–O– group. The $^1\text{H-}$ and $^{13}\text{C-NMR}$ data of compound **2** were found to similar to that of **1** except two more olefinic proton signals in $^1\text{H-NMR}$ at δ 5.02 (dd, $J = 15.0, 8.5\text{ Hz}$) and δ 5.15 (dd, $J = 15.0, 8.5\text{ Hz}$) were assigned for olefinic protons at C-22 and C-23 respectively. The coupling constant suggests that these protons are trans in position. From all these spectral analysis, the compound was identified as stigmasterol.

Compound **3** was obtained as white amorphous powder having melting point 284°C. The mass spectrum showed the M^+ peak at m/z 576 corresponding to molecular formula $C_{35}H_{60}O_6$. The IR spectrum showed the absorption of

hydroxyl groups 3400 cm^{-1} and –C–O group at 1063 cm^{-1} . The $^1\text{H-}$ and $^{13}\text{C-NMR}$ data of compound **3** were found to similar to that of **1** except the signals of glycoside moiety resonate at δ 3.96–5.02. A signal at δ 5.02 was assigned for the anomeric proton of glucose moiety attached at C-3. Other multiplet at δ 4.27 was assigned for the proton of C-3. Its deshielding may be due to attachment of $\hat{\alpha}$ -O-glucosyl moiety at C-3 carbon. All the spectral values obtained from $^1\text{H-NMR}$ and $^{13}\text{C-NMR}$ and comparing with authentic compound the compound was identified as $\hat{\alpha}$ -sitosterol glucoside (**8**).

Compound **4** was isolated as white crystalline solid, m. p. 75°C. The EI-MS mass spectrum showed (m/z 424, M^+) corresponding to molecular formula $C_{29}H_{60}O$. IR spectrum gave hydroxyl group at 3350 cm^{-1} and C–O stretching at 1200 cm^{-1} . The peak at 2992 cm^{-1} showed that the compound was aliphatic. The $^1\text{H-NMR}$ spectrum showed a triplet equivalent to three protons at δ 0.88 ($J = 7.0\text{ Hz}$) which were assigned as two terminal methyl group and a broad singlet equivalent to 34 protons at δ 1.28 were assigned for 17 methylene groups in an aliphatic chain. The multiplet at δ 3.60 was assigned to the methane proton of the carbon containing hydroxyl group. All these evidences were in agreement with the compound being nonacosan-10-ol (**8**).

CONCLUSION

Tsuga dumosa D. Don is an economically as well as medicinally important Himalayan conifer. The Phytochemical screening and isolation of chemical constituents of *T. dumosa* afforded three sterols, namely $\hat{\alpha}$ -sitosterol (**1**), stigmasterol (**2**), $\hat{\alpha}$ -sitosterol glucoside (**3**) and an alcohol, nonacosan-10-ol (**4**). The structures of the isolated compounds were deduced from the analysis of spectroscopic data as well as with comparing with the authentic samples.

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