

Application of Multivariate Approaches to the Hydro-chemical Assessment of the Ghodaghodi Lake, Sudurpaschim Province, Nepal

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ABSTRACT

Wetlands are considered the most diverse and productive ecosystems in the world, providing life-sustaining ecological services. Hydrochemical variables of the Ghodaghodi Lake situated in Far west Nepal were studied using multivariate statistical approaches to investigate the characteristics of water quality of the lake. A total of 10 water samples were collected and analyzed for water temperature, pH, electrical conductivity, total dissolved solids, total hardness, salinity, dissolved oxygen, major ions, and dissolved Si in the pre-monsoon season during 2017. The water quality assessment was carried out by applying sodium percentage, sodium adsorption ratio, permeability index, Kelly's ratio, magnesium adsorption ratio, action ratio of soil structural stability, and water quality index. Multivariate statistical techniques were used to evaluate the geochemical and anthropogenic processes and to identify factors influencing the ionic concentrations. The results revealed that the water was slightly alkaline with low ionic strength and remarkable spatial variations. The hydrochemistry of the lake was mainly controlled by rock weathering as the dominant $\text{Ca}^{2+}\text{-HCO}_3^-$ facies. The principal component analysis provided three major components exhibiting the diverse sources of natural and anthropogenic chemicals, including agricultural fertilizers, leakage of sewages near the human settlements, etc. The measured hydrochemical parameters indicate that lake water lies within the safe drinking water and irrigational standards. However, special consideration should be taken to control the concentrations of NO_3^- due to increasing anthropic activities.

Keywords: Hydrochemistry, statistical analysis, water quality

1. INTRODUCTION

Wetlands are one of the primary and most trusted sources of freshwater in the world (Khadka & Ramanathan 2013). The wetland ecosystems are the most diverse, providing life-sustaining ecological services and economic values (Pant & Adhikari 2015). Water purification, nutrient retention, groundwater recharge, biodiversity conservation, food supply, etc., are critical functions directly associated with the wetland ecosystem (Singh *et al.* 2016). However, during the last few decades, productive wetland eco system suffer from intense anthropogenic pressure (Pant *et al.* 2019a). The most important natural surface processes that regulate freshwater bodies' hydrochemistry include precipitation, geochemical weathering, recharge and discharge, ion exchange and redox reactions, and evapo-crystallization (Gibbs 1970). Anthropogenic activities, including leaching of fertilizers from farmlands, industrial effluents, and domestic wastes, can also affect water quality in the lakes (Pant *et al.* 2019).

The freshwater hydrochemistry describes the interactions between exogenic and endogenic processes, the behavior of dissolved chemical variables, and watershed characteristics. The freshwater lakes constitute a unique geochemical process as they are less dynamic and more susceptible to both natural and anthropic interventions (Rupakheti *et al.* 2017). The chemical quality of the lakes has been widely controlled by weathering of underlying rocks, evapo-crystallization, and atmospheric precipitation in the natural conditions whereas, the anthropic inputs such as urban runoff, industrial effluents, and land use/cover change are the major sources of contaminants (Pant *et al.* 2018). In addition, global climate change has further aggravated the impacts on the quality and the quantity of the water resources, hence the study of freshwater quality has immense significance for the sustainability of human society and ecosystems.

Multivariate statistical analysis and geochemical indices are widely used to evaluate the hydrochemistry and pollution levels of the water bodies (Pant *et al.* 2018). Pollution in the wetland ecosystem is common in developing and developed countries and one of the serious environmental issues under global climate change. The chemical contamination in freshwater bodies is one of the severe issues from the Himalayas, especially from human consumption, irrigation, industrial and other purposes (Sharma *et al.* 2020; Pal *et al.* 2021). The contamination of freshwater bodies can result in low drinking and irrigation water quality and cause potential human and ecological health problems. In general, the fluctuation in the parameters, including ionic constituents in water beyond the permissible level due to various natural and anthropogenic processes (Pal *et al.* 2019; Pant *et al.* 2021a)

Although Nepal is rich in freshwater resources, the Ramsar sites and other freshwater lakes are under high pressure due to various anthropogenic activities and global climate change (Zhang *et al.* 2019). Overdependence on the water to meet the ever-increasing demands of domestic, agricultural, industrial, and haphazard developmental activities has resulted in the overexploitation of water resources and deteriorated lake water quality in Nepal (Rupakheti *et al.* 2017). Several studies were conducted in different parts of the country to evaluate the water quality and its suitability for various purposes, spatiotemporal distribution, surface water interactions, and water vulnerability (Khadka & Ramanathan 2013; Pal *et al.* 2019; Bishwakarma *et al.* 2019; Pant *et al.* 2013; Pant *et al.* 2018; Sharma *et al.* 2020; Pant *et al.* 2021b). These studies have highlighted the anthropogenic pressures from agricultural lands, urban and domestic sewage discharge, and the elevated chemical attributes pose safety apprehensions for drinking and irrigation purposes of freshwater bodies of Nepal when compared to WHO, US EPA, FAO, and other standard guidelines of water quality (WHO 2011).

Ghodaghodi Lake is one of the Ramsar listed (Ramsar site No. 1314) wetlands and the famous tourist destinations in the Sudurpashim Province, Nepal. The lake supports the livelihoods of local communities and is a staging area for migrating birds. Studies with regard to Ghodaghodi Lake Area is scattered and there are limited studies for the quality assessment of the current status of the lake. In this study, the multivariate statistical approaches, combined with geochemical indices, were adopted to characterize the water quality of the Ghodaghodi Lake. The findings of this study could be useful for Provincial and Local government and relevant stakeholders working in the area for the sustainable management of the Ghodaghodi Lake as an Important Ramsar sites of Nepal.

2. MATERIALS AND METHODS

2.1 Study Area

Ghodaghodi Lake is enlisted as Ramsar site as one of the largest inter-connected natural lake systems in Nepal's Terai region. It is located at the latitude of 28°41'17"N and the longitude of 80°56'47"E at 205m above sea level in the Ghodaghodi Municipality of Kailali District, Sudurpaschim Province, Nepal (Fig. 1). The maximum length of the lake is 1.83km and the average depth of lake is about 4m. Three small settlements surround the study area: Sandepani, Darakh, and Ramshikharjhala in the east, south, and north-west, respectively, and approximately covers an area of 2500ha. It has 14 large and small oxbow lakes, including main Ghodaghodi (138ha), Nakharodi (70ha), and

Baishawa (10ha), and are associated with marshes, swamps, streams, and springs, together with canals, irrigated fields, and ponds (Bhatta *et al.* 2020). The area is also surrounded by tropical deciduous mixed Salforest (*Shorea robusta*) in the Siwalik Hills' lower slopes. The lake system located inside the Basanta Protection Forests and is a biological corridor for mega fauna, the corridor connecting Bardia National Park in the east and Suklaphanta National Park in the west. The lake experiences a tropical monsoon climate, dry winter, and rainy summer. The maximum temperature of this area ranges from 22°C to 37°C and minimum from 5°C to 25°C. The average annual rainfall ranges between 1630mm and 1705mm, where about 80-85% of the total rainfall occurs during the monsoon season from June to late September (Lamsal *et al.* 2014).

2.2 Sample Collection and Analysis

The field work of this study was conducted in March 2017, and the sampling was carried out to allow maximum representativeness of the lake. The water samples were collected from 10 different sites of the lake (Fig.1). The representative samples were taken from the periphery and center on the basis of potential pollution sources, land use pattern of the surrounding areas, and disturbed and relatively less disturbed site. Water temperature, pH, EC, and TDS were measured on-site using a multi-purpose probe (YSI 6920). Besides, DO and salinity was also measured *in-situ* using the DO and salinity meters. For the collection of the samples, previously acid-washed 1L high-density polyethylene bottles were used.

All the samples were filtered through 0.45µm millipore nitrocellulose paper, and a small portion of samples was collected for measuring Cl^- , NO_3^- and SO_4^{2-} concentrations, while the remaining portion was acidified with ultra-pure HNO_3 to reduce the pH <2 for Na^+ , K^+ , Ca^{2+} , Mg^{2+} and Si determination as described by (APHA, 2005). During the laboratory analyses, distilled deionized water was used, and freshly prepared standards of known concentrations and procedural blanks were analyzed during each analytical run, and no detectable contaminations were found. Furthermore, each calibration curve was evaluated by analyses of a set of samples. The data for this study was collected for only one time which is one of the limitations of the study.

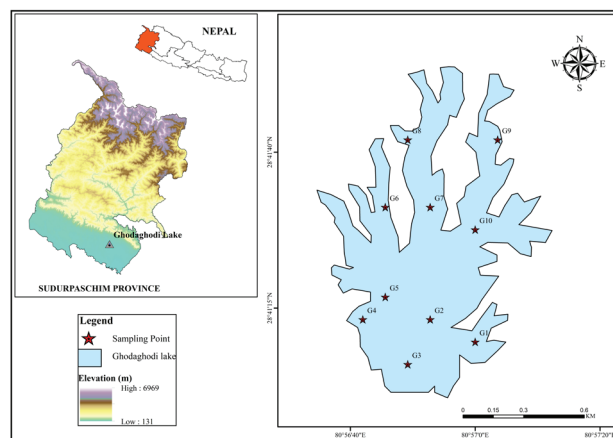


Fig. 1. Sampling locations in the Ghodaghodi Lake, Sudurpaschim Province, Nepal

2.3 HYDROCHEMICAL ANALYSIS

The study has intensively examined the major physicochemical parameters of both the on-site and laboratory as per the standard methods (APHA 2005). Proper care was taken to make the container neat and clean, avoiding impurities' effects in the results before sampling. A total of 16 parameters were analyzed, including significant cations and anions. Water temperature, pH, electrical conductivity (EC), salinity (SA), and total dissolved solids (TDS) were measured at the sampling site using a YSI 6920. The dissolved oxygen (DO) was determined by DO meter on the spot, and the total HCO_3^- was determined by titration with HCl on the same day. Major cations (Na^+ , K^+ , Ca^{2+} , and Mg^{2+}) and Si were determined using Inductively Coupled Plasma Optical Emission Spectrometer (ICP-OES, Prodigy). Anions (Cl^- , NO_3^- , and SO_4^{2-}) were measured using an Ion Chromatography of Dionex (ICS 900, USA) and all the laboratory analysis was conducted at the Institute of Tibetan Plateau Research, Chinese Academy of Sciences (ITP-CAS), Beijing, China. Each calibration curve was evaluated by analyzing quality control standards before, during, and after the samples' analysis. Geochemical characterization of water quality in the lake was assessed by the Piper (1944) diagram, whereas Gibbs (1970) plots did the controlling mechanisms of natural water quality.

2.4 Irrigation Suitability Assessment

The chemical and physical characteristics of the lake water are the fundamental consideration for the irrigation water quality evaluation. The specific hydrochemical parameters such as EC, HCO_3^- , Ca^{2+} , Mg^{2+} , Na^+ , and K^+ are considered

to analyze irrigation water quality. Such properties have relevant relation to the growth and productivity of crops. The irrigation water quality of different sections in the Ghodaghodi Lake has been applying EC, sodium percentage (Na %), sodium adsorption ratio (SAR), magnesium adsorption ratio (MAR), Kelly's ratio (KR), permeability index (PI), cation ratio of soil structural stability (CROSS), and Wilcox diagram. Also, the lake's drinking water quality was evaluated using a water quality index (WQI). The detailed methodologies for irrigational quality assessment are explained elsewhere (Acharya *et al.* 2020).

2.5 Data Analysis

Principal component analysis (PCA) is one of the most important multivariate statistical tool, identifying the various processes that control the solute concentrations (Singh *et al.* 2004; Filzmoser *et al.* 2009). PCA was computed for the EC, TDS, Ca^{2+} , Mg^{2+} , Na^+ , K^+ , Si, Cl, NO_3^- , SO_4^{2-} and HCO_3^- to identify the sources of different hydrochemical parameters. The KMO-test was carried out to examine the data set's suitability, and factor loadings were determined after plotting the screen plot. The extraction factors were based on the variances and covariance of the variables. The factor with eigen values >1 is considered as the significant variable in this study. Finally, by the process of rotation, each variable's loading on one of the extracted factors is maximized, and the loadings of all the other factors are minimized. For the statistical analysis, the software package (SPSS 26) was used (Sajil Kumar 2016).

3. RESULTS AND DISCUSSION

3.1 General Hydrochemistry and Spatial Distribution

A descriptive statistical summary of major hydrochemical compositions of the Ghodaghodi Lake is presented in Table 1. The lake water was mildly alkaline, pH ranging from 7.11 to 8.80 with a mean value of 7.96 ± 1.54 . The lake's dissolved oxygen (DO) was observed between 3.14 mg/L to 7.41 mg/L, with a mean value of $5.54 \text{ mg/L} \pm 1.35$ mg/L, indicating the aquatic species can survive in the lake (Gautam *et al.* 2015). Lake water has low TDS, ranges between 63 to 98 mg/L, and low mineralization, as shown by the EC measurements, varying from 112 to 182 $\mu\text{S/cm}$. Ca^{2+} and Na^+ are the most abundant cations with average concentrations of 13.96 mg/L to 17.11 mg/L and 4.24 mg/L to 9.26 mg/L, respectively. K^+ is the least abundant major cation with an average concentration of 1.46 mg/L to 3.09

mg/L. The cationic dominance was found in the order of $\text{Ca}^{2+} > \text{Na}^+ > \text{Mg}^{2+} > \text{K}^+$.

HCO_3^- is the most abundant major anion, ranged from 35 mg/L to 65 mg/L. The second most abundant anion is SO_4^{2-} ranging from 3.25 mg/L to 6.03 mg/L. The order of the dominance of anions is $\text{HCO}_3^- > \text{SO}_4^{2-} > \text{Cl}^- > \text{NO}_3^-$. Dissolved silica has a lower concentration with a mean value of 1.08 mg/L. Among the major ions, the highest mean concentration was found for the HCO_3^- . By comparing with the World Health Organization (WHO 2011) for drinking water (Table 1), all the hydro-chemical variables are within the permissible limits for drinking water in the Ghodaghodi Lake. Also, excessive loading of NO_3^- contributes to Lake Eutrophication. The mean value of NO_3^- is >2 mg/L. Thus, there is an indication of possible enrichment of NO_3^- but it is still within the permissible drinking water limits.

The distinct spatial variations were observed in the mean concentrations of measured hydrochemical variables in the Ghodaghodi Lake (Fig. 2). For instance, the pH, EC, TDS, Ca^{2+} , Mg^{2+} , and HCO_3^- were relatively higher in the outlet sections of the lake (i.e. sampling No. G1 to G5, Fig. 1). On the other hand, the mean concentration of dissolved Si and SO_4^{2-} were relatively higher in the inlet section of the lake. Interestingly, some of the sampling locations such as G1 and G6 are characterized by the elevated concentration of the K^+ . Whereas, NO_3^- was markedly higher in the G10. The spatial variations of hydrochemical parameters indicated that both the point and non-points sources of pollution are prevalent in the lake basin. Meanwhile, the inlet section of the lake basin is relatively less contaminated as compared to the outlet section except for G10 for NO_3^- . In the proximity of G10, intense farming has been existing thus the major sources of the NO_3^- could be agricultural activities. Finally, the outlets' section (G1) is characterized by the elevated concentration of most of the chemical variables which could be due to the anthropic activities in the nearby settlements.

Table 1: Descriptive statistics of the physicochemical composition of the Ghodaghodi Lake, Sudurpaschim Province, Nepal

Parameter	Min	Max	Mean	SD	WHO drinking water limit
Temp.	30	34.7	32	1.54	-
pH	7.1	8.8	8	0.64	6.5-8.5
EC	112	181	142	26.52	1500
TDS	63	98	77	13.69	1000
SA	28	40	31	3.52	-
DO	3.2	7.41	5.5	1.35	5
Ca^{2+}	14	17.11	16	1.034	100
Mg^{2+}	2.3	2.67	2.4	0.12	50

K ⁺	1.5	3.09	2.1	0.54	100
Na ⁺	4.2	9.26	5.5	1.48	200
Si	0.7	1.5	1.1	0.31	-
Cl ⁻	4.3	8.7	6.6	1.51	250
NO ₃ ⁻	1.7	3.61	2.1	0.56	50
SO ₄ ²⁻	3.3	6.03	4.8	0.94	250
HCO ₃ ⁻	35	65	49	10.58	600

The temperature in °C and other concentrations are expressed in mg/L except for EC in (µS/cm) and pH

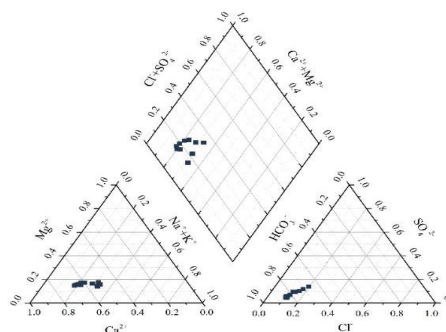


Fig. 3. Piper diagram showing the Ca²⁺-HCO₃⁻ hydrochemical facies of Ghodaghodi Lake, Sudurpaschim Province, Nepal

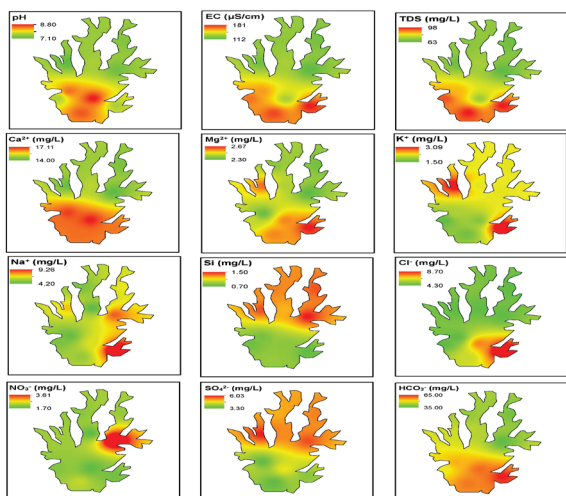


Fig. 2 Spatial variations of the physicochemical parameter in the Ghodaghodi Lake, Sudurpaschim Province, Nepal

3.2 Characterisation of Hydrochemical Facies

The lake’s water type was identified through the Piper trilinear diagram (Fig. 3). The piper diagram reveals that the dominance of a weak acid (HCO₃⁻) over strong acid (SO₄²⁻ + Cl⁻) and alkaline earth metals (Ca²⁺ + Mg²⁺) over alkali metals (Na⁺ + K⁺). It shows that Ca²⁺-HCO₃⁻ is the central dominant hydrochemical facies in the Ghodaghodi Lake. Major cations and anions’ arrangements suggested that the freshness of these water samples with carbonate weathering as a dominant mechanism regulating the lake’s hydrochemistry. The hydrochemical facies of the Ghodaghodi Lake is also comparable with the other lakes of Nepal Himalaya such as Phewa Lake, Jhilmila Lake, Begnas Lake, Rupa Lake, etc. (Pant & Adhikari 2015, Pant *et al.* 2019, Adhikari *et al.* 2020; Pal *et al.* 2021). The dominance of Ca²⁺ and HCO₃⁻ in total cation, and anion, respectively in the Nepal Himalaya region is mainly due to the intense carbonate rock weathering (Pant *et al.* 2018).

3.3 Source Appointment and Controlling Factors

3.3.1 Principal Component Analysis (PCA)

Based on the 11 hydro chemical variables, PCA analysis was applied to identify the variables with the most significant explanatory power that govern the lake water’s chemical composition (Adhikari *et al.* 2020). The results of principal components (PCs), variable loadings, and variance are presented in Fig. 4. The highest eigen values are the most significant, and eigen values >1 are considered significant (Li & Zhang 2010). The component loadings are ‘strong’ and ‘moderate,’ corresponding to absolute loading values of >0.75 and 0.75-0.50. Three powerful PCs have observed eigen value > 1.0 in this study.

The PC1 accounts for 56.88% of total variance with strongly favorable loading on EC, TDS, Ca²⁺, and HCO₃⁻; moderate positive loadings on Cl⁻, whereas strong negative loadings on Si and SO₄²⁻. The possible sources of constituents in PC1 are natural ones, mainly derived from carbonate rocks’ weathering. PC2 is responsible for 23.59 % of total variance with strong positive loadings on K⁺ and Na⁺ and moderate positive loadings on Mg²⁺ and Cl⁻, suggesting the silicate weathering. PC3 accounts for 9.77 % of total variance with strong loadings on NO₃⁻. The contamination of NO₃⁻ in lake water mainly from anthropogenic activities- is one of this study’s key findings. Most of the lake’s area vicinity is used for agricultural and religious purpose. Excessive use of fertilizers and organic chemicals could be the NO₃⁻ sources in the lake water. Also, NO₃⁻ in the surface water is contributed from the decomposition of soil organic matter, leaching of nitrogen-containing chemical fertilizers, and animal excreta. The results revealed that the sampling points nearby the settlements and intensive farming are relatively more contaminated as compared to the central section of the lake. These results are consistent with the spatial variations (Fig. 2) as it was observed that the diverse sources of natural and anthropogenic chemicals, including agricultural fertilizers, leakage of sewages near the human settlements, etc. could be the major sources of contamination.

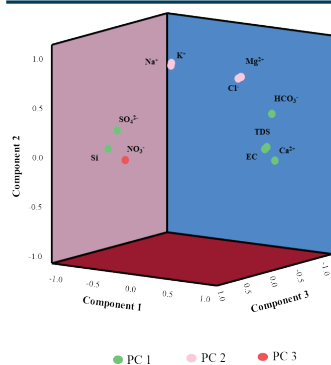


Fig. 4. Factor loading plot of the principal component analysis for EC, TDS, and major ions

in the Ghodaghodi Lake, Sudurpaschim Province, Nepal

3.3.2 Gibbs Plots

The lake's chemical substances have multiple sources, such as weathering of carbonate, silicate, evaporites (the chemicals including sodium salts, and halite that crystallizes from evaporating lake water), and sulfide minerals, including anthropogenic contributions and cyclic salts (Varol *et al.* 2013). Gibbs (1970) provided a simple plot of TDS versus the weight ratio of $\text{Na}^+(\text{Na}^+\text{+Ca}^{2+})$ and $\text{Cl}/(\text{Cl}+\text{HCO}_3^-)$ to determine the relative importance of three major-natural mechanisms including atmospheric precipitation, evapo-crystallization, and rock weathering that control the surface lake water chemistry. The TDS concentration and the weight ratios of $\text{Na}^+(\text{Na}^+\text{+Ca}^{2+})$ and $\text{Cl}/(\text{Cl}+\text{HCO}_3^-)$ plotted for the Ghodaghodi Lake revealed that the lake water has a low ratio of $\text{Na}^+(\text{Na}^+\text{+Ca}^{2+})$ and $\text{Cl}/(\text{Cl}+\text{HCO}_3^-)$, and moderate TDS concentrations (Fig. 5). It has indicated the domination of rock-weathering as a primary controlling mechanism in the hydrochemistry of the lake. The rock dominance in the lake basin provides insight into chemical weathering in the drainage since weathering of different parent rocks (e.g., carbonates, silicates, & evaporites) yields different combinations of dissolved cations and anions to the lake water. For instance, Ca^{2+} and Mg^{2+} originate from the weathering of carbonates, silicates, and evaporates; Na^+ and K^+ from the weathering of evaporites and silicates; HCO_3^- from carbonates and silicates, while SO_4^{2-} and Cl^- from evaporites, and silica exclusively sources from the weathering of silicates (Chen *et al.* 2002). The lake's ion chemistry can also be affected by human activities (Meybeck & Helmer 1989; Singh *et al.* 2016). The anthropogenic impacts have been increasing recently due to the dramatic increases in industrial, agricultural, urbanization, and domestic activities. Though major-ion compositions on the lake are within the range of natural water chemistry, the signature of elevated concentrations of Cl^- , NO_3^- and SO_4^{2-} in the lake samples

are due to anthropogenic sources. Notably, NO_3^- has significantly higher concentrations (>2 mg/L). The results also highlight the signature of religious activities around the lake.

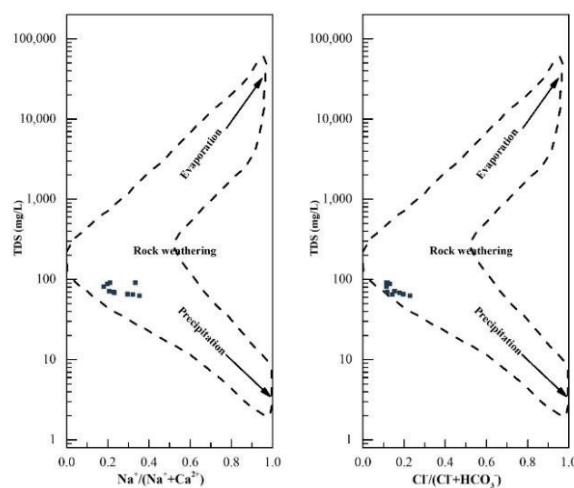


Fig. 5. Gibbs plot showing rock weathering is the primary natural controlling mechanism of hydrochemistry in the Ghodaghodi Lake, Sudurpaschim Province, Nepal

3.4 Water Quality for Irrigation and Drinking Purpose

The Ghodaghodi Lake water irrigation suitability has been evaluated using different methods and presented in Table 2. EC's classification suggested that all the points are in the "Excellent" category and concluded that the lake water is acceptable for irrigation purposes. About 21,000 ha area of the Ghodaghodi Municipality is used for the agricultural purposes. The majority of the area has been irrigated by the water from the lake complex. $\text{Na}\%$ in the lake ranges from 18.49% to 32.80%, categorized as Good and Excellent. Additionally, the SAR values of this lake water (<0.60) indicated the "Excellent" category and signify the potentiality of water for irrigation purposes. Both the MAR (<30) and KR (<0.50) values belong to the safe category for irrigational usage, suggesting that lake water is suitable for irrigation (Acharya *et al.* 2020)Nepal. In this study, the surface water samples were collected from 16 sites with a 0.5 km interval to characterize and evaluate the water quality mainly from the perspective of its irrigational usage. The assessment was carried out by applying electrical conductivity (EC). The PI's value was found to be <115 displayed the lake lies in category I, i.e., good. The CROSS values of the lake are found in the "Excellent" category. Furthermore, lake water's suitability for drinking purposes was assessed using the WQI, and found that water is suitable for drinking purposes based on measured hydro chemical variables.

Table 2. Drinking and irrigational water quality of the Ghodaghodi Lake, Sudurpashchim Province, Nepal

Parameter	Average Value	Category
WQI	8.68	Excellent
EC	142	Excellent
Na%	24.77	Good
SAR	0.37	Excellent
MH	21.35	Suitable
KR	0.28	Safe
PI	103.89	Class I
CROSS	0.44	Excellent

4. CONCLUSION

The study was done for hydro chemical investigation in the Ghodaghodi Lake water by applying multivariate statistical approaches and geochemical indices. The lake water was mildly alkaline with relatively low EC and TDS values. The dominance order of major cations and anions were $\text{Ca}^{2+} > \text{Na}^+ > \text{Mg}^{2+} > \text{K}^+$, and $\text{HCO}_3^- > \text{SO}_4^{2-} > \text{Cl}^- > \text{NO}_3^-$, respectively. The characterization of hydrochemistry was carried out by plotting the Piper diagram, and only Ca^{2+} - HCO_3^- type of water facies was observed, which exhibited the carbonate-dominated underlying lithology in the lake basin. From the PCA, three principal components were derived, which explained >90% of the variations. PC1, PC2, and PC3 were high loadings with EC, TDS, Ca^{2+} , and HCO_3^- ; K^+ , Na^+ , and Cl^- ; and NO_3^- indicating the geogenic, mixed, and anthropogenic sources of origin, respectively. These results are also in good agreement with Gibb plots.

While comparing the hydro chemical variables with WHO drinking water guideline values, the lake water was safe for drinking water purposes. Similarly, irrigation water quality standards indicated no severe threats in the water quality currently, however increasing concentration of the variables may pose some issues in the future. Nevertheless, special attention should be given to control the increasing concentration of NO_3^- due to various anthropogenic stresses in the lake's vicinity. The research finding will help provide new insights into the sustainable water quality of the Ramsar listed wetlands. A further in-depth study focusing on hydro chemical variation concerning the lake's seasonality impacts is recommended for future study.

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