

An isoquinoline alkaloid, Protopine from *Corydalis govaniana* growing in Himalayan region of Nepal

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Abstract

Chemical investigation of the roots of *Corydalis govaniana* resulted in the isolation and characterization of one isoquinoline alkaloid, protopine(1). The structure was established on the basis of extensive spectroscopic data analysis and comparison with spectroscopic data reported.

Key words: *Corydalis govaniana*, Fumariaceae, Isoquinoline alkaloid.

Introduction

Corydalis govaniana Wall (Fumariaceae) is glabrous herb and is distributed in Himalayas of Pakistan, India and in Eastern, Central and Western part of Nepal. It grows in damp and shady places at 2400 to 4800 m attitude¹. A number of medicinal values have been reported in Indian and Chinese systems of medicine for corydalis species^{2,4}. The root of *Corydalis govaniana* is used during diarrhea and dysentery². Previous studies on *Corydalis govaniana* led to the isolation of many isoquinoline alkaloids^{5,6,7}. Owing to its unusual geographical location, it was decided to analyze the plants in terms of its alkaloidal content. Our study of a *Corydalis govaniana* has resulted in the isolation and characterization of one isoquinoline alkaloid (1).

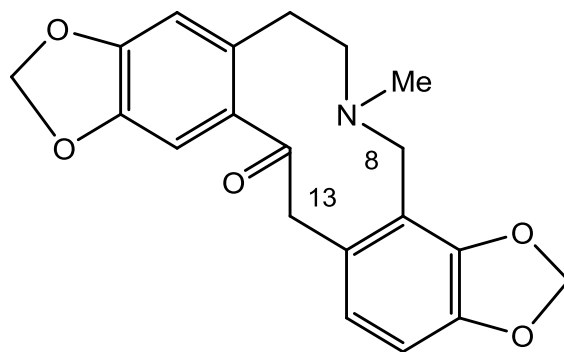


Figure 1: Protopine : 4,6,7,14-Tetrahydro-5-methylbis[1,3]benzodioxolo[4,5-c:5',6'-g]azecin-13(5H)-one

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Experimental

The melting point was determined on a Toshniwal apparatus and uncorrected UV spectrum was recorded with Perkin-Elmer Lambda spectrophotometer using spectral methanol. An IR spectrum was recorded in KBr pellets. ¹HNMR spectra were recorded in 500 MHz in deuterated acetone using tetra methyl silane (TMS) as internal reference. Mass spectrum was performed on JEOL mass spectrometers operating at 70 eV. The purity of substance was checked on TLC plates.

The roots of *Corydalis govaniiana* was collected from Langtang, Nepal and identified by comparison with the authentic herbarium specimen at the National Herbarium Laboratory, Kathmandu, Nepal. Air-dried roots of *Corydalis govaniiana* (2kg) was extracted with methanol for seven days in cold percolator. After removal of methanol under reduced pressure, the residue (354.30gm) was treated with 7% citric acid and separated to alkaloid fraction according to the procedure of R.N. Jha et.al.⁸

The fraction obtained using above procedure was analyzed on TLC for alkaloids by spraying with Dragendorff's reagent. The chloroform extract (41.12gm) was chromatographed over silica-gel column using solvent of increasing polarity. The eluants from C₆H₆: Acetone (1:1) were mixed and solvent removed to give crude product which was re-chromatographed over silica-gel column to get pure compound which on crystallization from methanol yield 20 mg of compound (**1**).

Compound 1: M.P. 207° - 208° C, uv λ max (MeOH) 293 mμ, IR (KBr, cm⁻¹) 1675, ¹HNMR (in Table 1), 2D NMR in figure 1(2), ms (m/z,%), 353(10)148(100) (Scheme -1)

Table: 1 500 MHz ¹HNMR spectral data of Compound **1** in Acetone

| Chemical shift | Proton count | Splitting pattern | Probable assignment |
|----------------|--------------|-------------------|-----------------------------------|
| 2.04-3.83 | 8H | <u>m</u> | C-5-H2, C-6-H2, C-8-H2, & C-13-H2 |
| 5.93 | 2H | <u>s</u> | CH ₂ O- |
| 5.97 | 2H | <u>s</u> | CH ₂ -O- |
| 6.91 | 1H | <u>s</u> | C-1-H |
| 6.63 | 1H | <u>s</u> | C-4-C-H |
| 6.72 | 1H | <u>d</u> | C-11-H |
| 6.78 | 1H | <u>d</u> | C-12-H |
| 1.88 | 3H | <u>s</u> | N-CH ₃ |

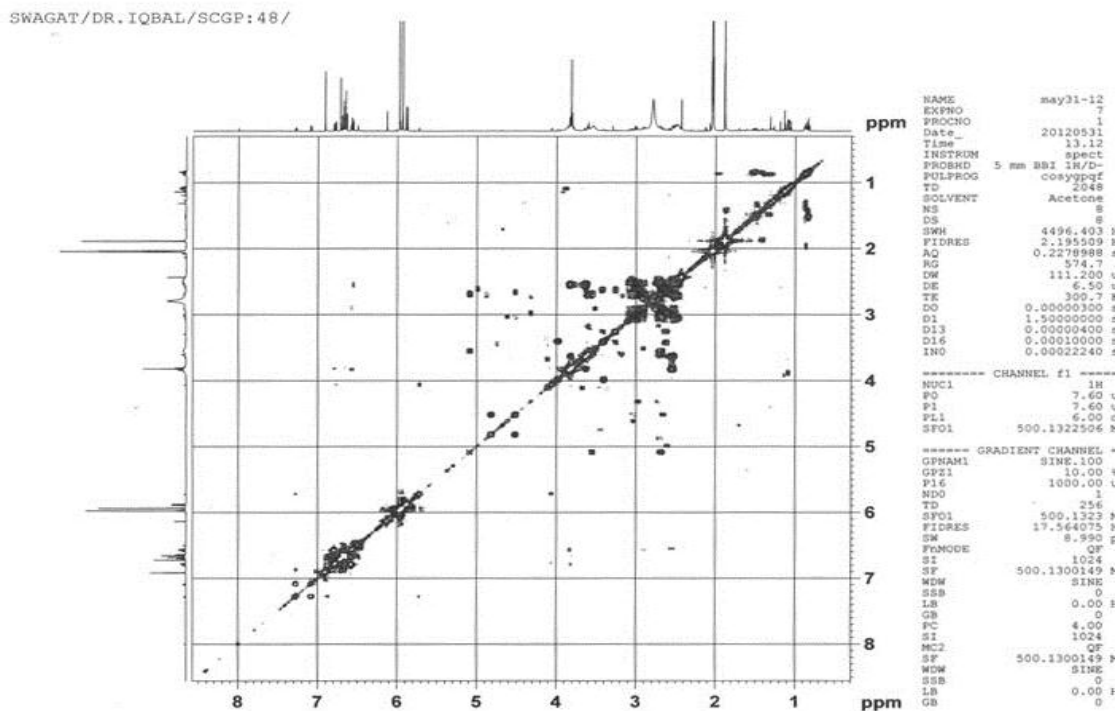


Figure 2: 2D NMR/COSY

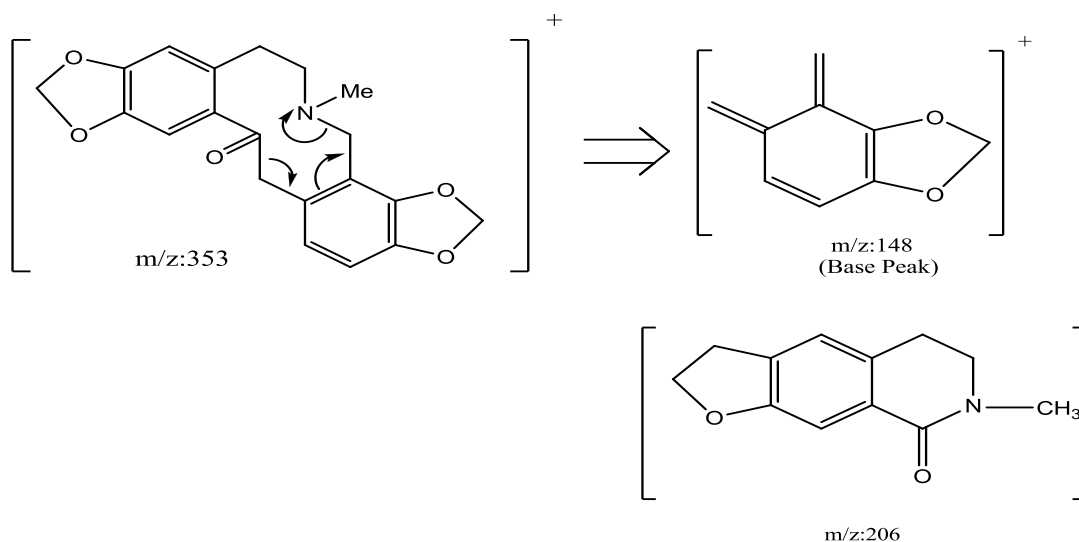
Result and Discussion

The compound (**1**) was obtained as colourless granules. The molecular formula of compound based on the high resolution mass spectrum was found to be $C_{20}H_{19}NO_5$; m/z 353 (M^+), 148 (base peak). A fragmentation pattern indicative of protopine could be easily rationalized (scheme 1). The ultraviolet spectrum in MeOH showed absorption maxima at 293 nm like that of isoquinoline alkaloids⁹. The IR spectrum contained an absorbance at 1675 cm^{-1} indicating the presence of a trans annular ground state interaction between the carbonyl and the basic nitrogen, so that the carbonyl group is appreciably of the amide type; $O=C \leftarrow N-CH_3$. The 1H NMR spectrum of **1** (Table-1) showed the following; two methylene dioxy group at δ 5.93 and 5.97 (each 2 H S); two aromatic protons as singlet at δ 6.91 and 6.63; two coupled aromatic protons as doublet at δ 6.78 and 6.72; one N-methyl group at δ 1.88(3H,S); four aliphatic protons as multiplet at 2.04 – 3.83. Its structure was also favored by 2D NMR. Based on above evidence, the structure of **1** was determined (Fig.1) and named as protopine. Further structure of this compound was confirmed by the comparison with reported data^{10,11}.

Conclusion

The structure of compound **1** was established by physical, chemical and spectroscopic data. The physical constant and spectroscopic data (IR, UV, NMR, MS) of the base were identical to those of protopine. This is the first report of the occurrence of protopine in *Corydalis govaniiana* growing in Himalayan region of Nepal.

Scheme 1



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